A parameterization of the specific surface area of seasonal snow for field use and for models of snowpack evolution

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[1] The specific surface area (SSA) of snow is needed to model air-snow exchange of chemical species. SSA is related to many snow physical properties, such as albedo and permeability. However, it is not described in models of snowpack evolution, in part because it is difficult to measure. Snowpack models often predict snow grain shape and snow density, and the goal of this paper is to propose parameterizations of snow SSA, based on snow density and grain shape. SSA values of 345 snow samples from snowpacks of the Alpine, maritime, tundra and taiga types are presented. Samples are regrouped into three main types: fresh (F), recent (R), and aged (A) snows, with several subtypes referring to grain shapes. Overall, there is a clear inverse correlation between SSA and density, d. Empirical equations of the form SSA = A ln(d) + B are proposed for the F and R types. For aged snows, separate correlations are proposed for subtypes A1 (rounded grains), A2 (faceted crystals), A3 (depth hoar), and A4 (lightly melted snow). Within subtypes A1, A2, and A3, more elaborate classifications are made by considering the snowpack type (Alpine, taiga, or tundra). For A1, A2, and A3 types, different trends are related to different intensities of wind action, which increases in the order taiga, Alpine, and tundra. We finally propose three parameterizations of snow SSA with increasing sophistication, by correlating SSA to snow type, then to snow type and density, and finally to snow type, density, and snowpack type.


1. Introduction

[2] The specific surface area (SSA) of snow is its surface area accessible to gases per unit mass [Legagneux et al., 2002]. SSA is a physical variable that is crucial to understand both the chemistry [Dominé and Shepson, 2002] and the physics [Legagneux et al., 2002] of the snowpack. Among many examples of its usefulness, many semivolatile chemical species such as persistent organic pollutants (POPs) are thought to adsorb onto snow surfaces [Daly and Wania, 2004; Herbert et al., 2005a, 2005b] and quantifying their amounts in snow and their exchanges with the atmosphere requires the knowledge of snow SSA and its variations. The albedo of snow has been shown to be related to grain size [Warren, 1982], but the definition of snow grain size is not simple [e.g., Aoki et al., 2000]. A potentially better description of grain size is its specific surface area, as already suggested by Warren [1982]. Domine et al. [2006] have recently shown that snow albedo in the IR could indeed be well correlated with snow SSA. Airflow through snow, and hence the efficiency of air-snow exchanges [Albert et al., 2002] is determined in part by the air permeability of snow that depends on air friction on the surface of snow crystals [Jordan et al., 1999]. Most models describing the air permeability of snow use grain size to predict its value, but it is likely that SSA would be a more appropriate variable.

[3] Snow SSA has seen little use in snow physics and chemistry, both in field studies and models, because its measurement is difficult and time consuming, the main reliable method established at present for routine use being methane adsorption at 77 K [Dominé et al., 2001; Legagneux et al., 2002]. It is also for the moment difficult to predict the evolution of snow SSA during metamorphism. Models have been proposed under isothermal conditions, on the basis of the concept of Ostwald ripening, but difficulties have been encountered because the normalized distribution of grain size is not at steady state in the seasonal snowpack, and the equations of Ostwald ripening therefore cannot be used in a simple manner [Legagneux et al., 2004; Legagneux and Domine, 2005]. Empirical equations have been proposed [Cabanes et al., 2002, 2003] but the relationship between these empirical rates and environmental variables such as temperature, temperature gradient and wind speed is not well understood, which limits their application. Recently, Flanner and Zender [2006] have proposed a physically based model of SSA evolution that
is valid under both isothermal and temperature gradient conditions. They approximate snow crystals as spheres, and use adjustable parameters to solve the difficulties encountered by Legagneux and Domine [2005]. Their sophisticated model is not very simple to use, however, and still requires experimental validation for many environmental conditions. In any case, given the present difficulty in parameterizing snow SSA in models, authors modeling the air-snow exchange of POPs adsorbed onto snow surfaces during the metamorphism of a snow layer used oversimplified approximations, such as a linear decrease in SSA with time [Daly and Wania, 2004].

[4] Understanding its time evolution is not the only SSA-related question snow studies have to face. Often, chemical analyses of a snowpack whose detailed history is unknown are performed, and the chemical interpretation requires the knowledge of snow SSA. For example, Herbert et al. [2005a, 2005b] tried to interpret the concentrations of persistent organic pollutants (POPs) in Arctic snow. They found a correlation between POP concentrations and snow density and concluded that this correlation was probably due to a correlation between SSA and density. It is therefore clear that a more detailed interpretation could have been made if these authors had used a reliable method to measure or estimate SSA. To assist in these types of studies, we wish to propose here a method to estimate SSA from snow variables that can be easily and rapidly determined in the field. An additional objective of this work is that contrary to the objectives of physically based approaches [Legagneux et al., 2004; Legagneux and Domine, 2005; Planer and Zender, 2006], we wish to propose a method that does not require the knowledge of the age of the snow sample.

[5] SSA is clearly determined by a large number of physical variables but following Legagneux et al. [2002] we propose to estimate its value from an examination of the snow crystal type and from a measurement of snow density. Besides its field use, this method can simply be incorporated in current models of snowpack evolution, such as CROCUS [Brun et al., 1992] and SNOWPACK [Lehning et al., 2002] that predict snow density and crystal type.

[6] Legagneux et al. [2002] presented a compilation of 176 SSA values, proposed a classification of snow types adapted to the determination of SSA, and gave a density-SSA correlation for each snow type, which allowed the estimation of snow SSA within 25 to 40% at the 1σ confidence level. Since then, we have increased the number of measurements to 345, significantly improving the statistical value of many correlations. Our previous compilation was based on studies of snow in the Alps and the Arctic, thus dealing with snowpacks of the tundra, Alpine and maritime types according to the classification of Sturm et al. [1995]. We have now performed many measurements in central Alaska, which allowed us to characterize the SSA of the taiga snowpack. We have also done measurements on snow layers affected by numerous melt-freeze cycles, which have very low SSAs. We further indicate correspondence between our snow type classification and that recommended by the International Classification of Seasonal Snow on the Ground (ICSSG) [Colbeck et al., 1990], although the latter one is not very well adapted to the description of SSA. Finally, we have detected and corrected an artifact in our previous protocol, which improves the reliability of our method.

2. Methods

[7] The method has been extensively detailed by Legagneux et al. [2002] and only the principle and recent modifications will be described. The principle is to measure the adsorption isotherm of methane on the snow sample at liquid nitrogen temperature, 77 K, using a volumetric method while snow is contained in stainless steel containers of volumes in the range 100–250 cm³. A Brunauer-Emmett-Teller (BET) mathematical treatment [Brunauer et al., 1938] is then applied to the isotherm to determine the surface area of the sample and the mean heat of adsorption of the first monolayer of CH₄ on ice surfaces. ∆QCH₄,snow The SSA is obtained by dividing the measured surface area of the sample by its mass.

[8] Since the compilation of Legagneux et al. [2002], an experimental artifact has been detected. The SSA of dry snow is usually in the range 80–1500 cm² g⁻¹, while the snow samples used weigh 15 to 50 grams. Thus the surface area of the sample is around 0.2 to 2 m². As briefly alluded to by Legagneux et al. [2004], we found that adsorption of CH₄ on the stainless steel walls of the vacuum container used for the measurement was not always negligible. We thus measured the adsorption isotherm of CH₄ on the walls of each container used and found that the surface areas of the containers were around 1000 cm² and that ∆QCH₄,stainlesssteel = 1300 J mol⁻¹. We then reinterpret the molar budgets of the volumetric measurements in terms of adsorption on both the snow and the stainless steel. Using this data analysis method, the snow SSA values were usually reduced by 6 to 12% with the containers used by Legagneux et al. [2002]. In our studies in Alaska, we used smaller containers to improve pumping speed during the measurements. This reduced the surface ratio between the snow and the stainless steel, and the correction, while still less than 10% for most samples, reached 50% for samples with a surface area less than 0.2 m², which concerns mostly snow having undergone melt-freeze cycles and aged depth hoar.

[9] Because ∆QCH₄,stainlesssteel is much lower than ∆QCH₄,snow uncorrected measurements yield a lower value of ∆QCH₄,snow. This effect is of course more important for samples with a low surface area, and without correction, ∆QCH₄,snow is an increasing function of the surface area of the snow sample, tending asymptotically toward the true ∆QCH₄,snow value, as observed by Dominé et al. [2000]. After correction, the correlation between ∆QCH₄,snow and surface area disappears, and there is just more scatter of ∆QCH₄,snow values at low surface areas, around a mean value ∆QCH₄,snow ± 1σ = 2540 ± 200 J mol⁻¹. Legagneux et al. [2002] proposed that the value of ∆QCH₄,snow be used as a test for the reliability of the measurement, and recommended ∆QCH₄,snow = 2240 ± 200 J mol⁻¹. After the discovery of this new artifact, we now recommend the value ∆QCH₄,snow = 2540 ± 200 J mol⁻¹.

[10] There is still a significant scatter in ∆QCH₄,snow values. A lot of our measurements were focused on the monitoring of the SSA of identified snow layers, either in the field [Cabanes et al., 2002, 2003] or in the laboratory
Legagneux et al. [2003, 2004] and we observed that some layers had, after correction, systematically low values, while others had high values. The scatter within a given layer whose SSA was measured 5–15 times over a few months was usually 60 J mol⁻¹, and we therefore suspect that variations in $\Delta Q_{\text{CH4,snow}}$ may be caused by different snow chemical compositions, with a fraction of the impurities being in the adsorbed state. Another possibility is that different snow samples have different predominant crystallographic faces, with different $\Delta Q_{\text{CH4,snow}}$ values, but this is not supported by the work of Chaix et al. [1996] and Chaix and Domine [1997]. A last possibility is of course experimental error, and in particular variations in the configuration of the volumetric system. We indeed observed that after rebuilding the system and changing various containers, the average $\Delta Q_{\text{CH4,snow}}$ could be modified by as much as 100 J mol⁻¹.

We believe that this is due to error in the measurement of the various volumes, to which $\Delta Q_{\text{CH4,snow}}$ is very sensitive. In conclusion, there seems to be several causes that can explain the large variability of $\Delta Q_{\text{CH4,snow}}$, but our data suggest that for a given configuration of a measurement system, the standard deviation in $\Delta Q_{\text{CH4,snow}}$ should be less than 100 J mol⁻¹.

[11] We found an excellent linear correlation between uncorrected and corrected surface areas of the sample ($R^2 = 0.999990$): $\text{SA}_{\text{corrected}} = 1.0014 \times \text{SA}_{\text{uncorrected}} - 902 \text{ cm}^2$. Hence we deduced that an excellent approximation of the correction is to subtract 0.9 times the surface area of the stainless steel container from the uncorrected snow surface area to obtain what we believe is the correct value. The factor 0.9 is caused by the less energetic adsorption of CH₄ on stainless steel than on ice.

[12] The reproducibility of the method, obtained by successive SSA measurements by two experimentalists studying several snow samples, as detailed by Legagneux et al. [2002] is $2\sigma = 6\%$. Including an estimate of systematic errors, the overall method accuracy is 12% [Legagneux et al., 2002].

3. Results and Discussion

3.1. SSA Values

[13] We present here 345 SSA values from seasonal snowpacks. The reproducibility of the method, obtained by successive SSA measurements by two experimentalists studying several snow samples, as detailed by Legagneux et al. [2002] is $2\sigma = 6\%$. Including an estimate of systematic errors, the overall method accuracy is 12% [Legagneux et al., 2002].

[14] 1. One-hundred-five SSA values are from the French Alps sampled at elevations between 1320 and 2890 m, in snowpacks of the maritime and Alpine types.

[15] 2. One-hundred-eleven samples are from the Arctic, sampled around Alert (Canadian high Arctic, 82.5°N, 67 samples), Barrow (Alaska, 71°N, 11 samples) and Ny Ålesund (Svalbard, 79°N, 33 samples). At Alert and Barrow, the snowpack is definitely of the tundra type. Near Ny Ålesund, the abundant precipitation due to the vicinity of the Gulf stream, combined with temperatures fairly warm for that latitude, often give the snowpack an Alpine character with thin depth hoar layers, frequent ice layers and snowpack thicknesses exceeding 1 m. In coastal areas, severe wind erosion forms a snowpack closer to the tundra type. Most samples from Ny Ålesund are from these coastal areas, and for simplicity all our Svalbard samples are classified as being of the tundra type.

[16] 3. One-hundred-twenty-nine samples from central Alaska, sampled at the Large Animal Research Station of the University of Alaska Fairbanks (65°N, 164°W), as described in detail by Taillandier et al. [2006]. The snowpack in central Alaska is typical of the taiga type, characterized by the predominance of depth hoar and faceted crystals. Of these samples, 30 were obtained from snow that precipitated on tables, under which air circulation prevented the establishment of high temperature gradients. These experiments reproduce some of the variables to which permanent snow is subjected on ice cap: low temperatures and low temperature gradients. The low gradients on ice caps are the results of the thermal insulation provided by snow layers that accumulate over the years. However, densification caused by the weight of these layers is not reproduced here. Despite this difference, these experiments may yield interesting data that will be difficult to obtain on ice caps, as liquid nitrogen is required for our measurements. Here we observed that snow layers with small rounded grains formed rather than depth hoar, and these were very similar to those encountered in cold Alpine regions with high accumulation. They were also similar to small-grained layers observed in ice caps, and which form mostly in winter [Alley et al., 1997; Dang et al., 1997]. Although the evolution of these snow layers was not entirely natural, they were subjected to variable environmental conditions similar to those found in other natural settings and the trends in density and SSA evolution were similar to cold Alpine snow layers. We have thus decided to include them in this compilation. We have not, however, included data obtained on natural samples evolved in laboratory cold rooms, as described by Legagneux et al. [2003, 2004].

[17] All the values are reported in Table 1 to show the variability, and are grouped by types, as defined by Legagneux et al. [2002]. Typical pictures of this classification are reported by Legagneux et al. [2002]. These types have a lot of features in common with those of the International Classification of Seasonal Snow on the Ground [Colbeck et al., 1990]. However, we found that the ICSSG classification was not always well adapted to describing and understanding SSA. For example, in the ICSSG classification, type 1sd is for stellar dendritic crystals and class 1gp is for graupel. In practice, snowfalls with just stellar dendritic crystals are rare, and these are usually accompanied by crystals with various degrees of riming. In a given snow episode precipitation often alternates between lightly rimed crystals and graupel-like crystals. Sampling the pure 1sd type is therefore a rare opportunity. Another example is that when the air temperature is just above 0°C, the SSA of falling snow is noticeably reduced while the aspect of snow crystals is little affected and a specific type, F4, had to be defined to account for this class. The main purpose of this classification is to describe snow according to the characteristics that affect its SSA. Whenever possible, we have nevertheless indicated in Table 1 the correspondence between both our and the ICSSG classifications.

[18] Similarly to that of ICSSG, our classification is based on the habit of precipitating particles and/or on the metamorphic history of the crystals. F stands for fresh snow, sampled at the most 24 hours after the fall and before any visible transformation has taken place. This corresponds to
<table>
<thead>
<tr>
<th>Snow Type</th>
<th>Subtype Description</th>
<th>Subtype Code</th>
<th>ICSSG Type</th>
<th>Mean SSA ± σSSA</th>
<th>Mean Density ± σdens</th>
</tr>
</thead>
<tbody>
<tr>
<td>Freshly fallen snow F</td>
<td>dendritic, from lightly rimed to graupel</td>
<td>F1 1313, 1231, 1029, 1026, 1013, 961, 960, 943, 931, 917, 892, 885, 857, 846, 837, 672, 637, 635, 630, 613, 577, 561, 553</td>
<td>1sd, 1gp</td>
<td>849 ± 210</td>
<td>0.084 ± 0.052</td>
</tr>
<tr>
<td></td>
<td>small columns, bullet rosettes</td>
<td>F2 1401, 768, 734, 730, 656, 565</td>
<td>1el</td>
<td>809 ± 299</td>
<td>0.102 ± 0.041</td>
</tr>
<tr>
<td></td>
<td>small plates, needles and columns</td>
<td>F3 1558, 1171, 1100, 1012, 918, 887, 813, 801, 748, 746, 746, 717, 681, 655, 645, 612, 610, 607, 599, 585, 571, 565, 555, 526, 489, 479, 469, 439, 432, 424, 399, 394, 376, 367, 331</td>
<td>1cl, 1nd, 1pl, 1ir</td>
<td>658 ± 260</td>
<td>0.108 ± 0.026</td>
</tr>
<tr>
<td>Snow with air T &gt; 0°C</td>
<td>dendritic, variably rimed</td>
<td>R1 901, 760, 577, 537, 527, 513, 507, 497, 491, 441, 435, 432, 431, 430, 423, 421, 419, 416, 416, 416, 415, 414, 412, 410, 408, 381, 380, 379, 371, 334, 318, 315, 303, 228, 226</td>
<td>2dc, 2bk</td>
<td>503 ± 112</td>
<td>0.148 ± 0.021</td>
</tr>
<tr>
<td></td>
<td>columns, bullets rosettes</td>
<td>R2 730, 727, 687, 664, 654, 644, 572, 564, 524, 530, 516, 505, 501, 475, 434, 427, 421, 418, 411, 408, 381, 380, 379, 371, 334, 318, 315, 303, 228, 226</td>
<td>2dc, 2bk</td>
<td>469 ± 141</td>
<td>0.156 ± 0.054</td>
</tr>
</tbody>
</table>
ICSSG type 1. R stands for recent snow with recognizable particles and corresponds to ICSSG type 2. The shape of R-type particles has undergone visible transformations, but the habit of these particles when they precipitated can still be recognized. A stands for aged snow. The shapes of these crystals have undergone major changes since precipitation and the original crystal shapes cannot be recognized. The A type regroups ICSSG types 3, 4, 5, 6, 9we and 9mc. Types 9rc and 9sc would probably be included in our A type, but we have not measured the SSA of such snows. S stands for surface hoar and corresponds to ICSSG type 7sh. W stands for windborne snow, sampled while airborne or immediately after deposition. The W type has no equivalent in the ICSSG classification, although it is probably somewhat related to a 2bk crystals that would be heavily rounded. Sub types are defined, with F1 for dendritic crystals with all levels of riming, from unrimed to graupel, F2 for columns and bullet rosettes that form in cold (T < −23°C) conditions, etc. This is detailed in Table 1. Regarding surface hoar, Legagneux et al. [2002] had only studied feather-shaped crystals, classified as S1 type, thus leaving open the possibility of having other surface hoar subtypes as S2, S3, etc. Here we have also sampled cup-shaped surface hoar, but given the limited number of measurements, we prefer for the moment to regroup all surface hoar samples under a single type, S.

[19] The age of R-type crystals ranges from a few hours to a few weeks. The age of A-type crystals can be just a few days and up to several months. Indeed, a wind storm or very high temperature gradient metamorphism can transform a crystal beyond recognition in just a couple of days. On the contrary, in sheltered high accumulation area where temperature gradients remain low, the initial habit of precipitating crystals can be recognized after a month and these snow layers would then be of the R type. Similarly to the ICSSG classification, the aspect of the crystals is the main criterion, not crystal age.

[20] Table 1 includes another subtype not studied by Legagneux et al. [2002]: A5, for snow layers that have undergone extensive melting or numerous melt-freeze cycles. When frozen, this snow type is very hard and cohesive, while type A4 has undergone only slight melting and remains of moderate hardness when frozen.

### 3.2. SSA-Density Correlations

SSA is expected to correlate to some extent to density, as confirmed in Figure 1. Legagneux et al. [2002] proposed fits of SSA-density correlations of the form ln(SSA) = Ad + B, where d is density and A and B are adjustable parameters, because Narita [1971] had proposed this analytical form. However, several empirical fits with simple analytical expressions (exponential, power and linear laws) were tested, and equation (1), with SSA in cm² g⁻¹ and d in g cm⁻³ was found to have the best correlation coefficient (R² = 0.431):

\[
SSA = -308.2 \ln(d) - 206.0. \tag{1}
\]

Figure 1, however, shows that equation (1) alone is a poor predictor of SSA. The mean error \( E_{\text{mean}} \) at the 1σ confidence level, obtained using equation (1), and defined as

\[
E_{\text{mean}} = \frac{4}{n} \sum_{i} \left( \frac{SSA_{i} - \bar{SSA}}{SSA_{i} + \bar{SSA}} \right)^{2}, \tag{2}
\]

where subscripts \( m \) and \( eq \) refer to values measured and predicted by equation (1), respectively, is 62%. The \( E_{\text{mean}} \) value is fairly high and, as already detailed by Legagneux et al. [2002], better predictions are obtained if the snow type is also considered. Figure 2 illustrates the relationship between
density, SSA and snow type: the average SSA and density values, as well as standard deviations, are plotted for all snow subtypes. There is a clear decreasing trend in SSA as snow ages and density increases. Only the A5 type stands out of the main trend, because the dominant process in its evolution, melting, is different from that of the other types, and results in a considerably reduced mean SSA: 29 \text{cm}^2 \text{g}^{-1}.

Density and SSA could be measured together on only three samples of this type, and two of these were during snowmelt in Alaska. Water drained out of the permeable snow, which was originally depth hoar, and the density therefore remained low. We suspect that a wider range of measurements would show a greater range of densities for this type.

[24] The higher SSA values are obviously for fresh snow, and especially for dendritic snow. For dry fresh snow, the mean value is 741 ± 260 \text{cm}^2 \text{g}^{-1}. Since our measurement reproducibility is 6%, the variability observed is real and not the result of experimental error. Microscopic observations of fresh snow reveal that an important cause of variability is sublimation during precipitation. For example, dendritic crystals falling in warm air presumably with low relative humidity showed significant rounding and the disappearance of small structures, indicative of sublimation [Nelson,
This was not observed when crystals fell in air around −15°C [see, e.g., Cabanes et al., 2002, Figure 5]. Other causes of variability are illustrated by the two snowfalls composed of bullet rosettes studied by Cabanes et al. [2002, Figure 1], which had values (corrected for the artifact mentioned above) of 734 and 1401 cm² g⁻¹. The values are different because the snowfall with the higher SSA had much smaller crystals, which can be due to a higher number of ice nuclei in the cloud and to a lower moisture content limiting growth. In summary, differences in SSA of fresh snow can be caused by a large number of factors originating in the cloud or during precipitation. At this stage, we cannot propose a prediction of the SSA value of falling snow as a function of meteorological parameters, but we anticipate that further studies may offer predictions of the SSA of falling snow as a function of ice nuclei concentrations, cooling rate, available moisture, and the temperature profile of the atmosphere. At present, the best guess we can propose for dry fresh snow is the average value mentioned above: 741 cm² g⁻¹. As shown in Table 1 and Figure 2, considering the type of crystals allows extra precision.

Figure 3. SSA-density correlation for fresh snows. Although subtypes F1 to F4 are located at different places in the graph, they follow a similar trend. Recently wind-blown snow, W1, also follows this trend, and a single equation is proposed to parameterize SSA from the density for all five subtypes.

Figure 4. SSA-density correlation for recent snows. Subtypes R1 to R4 are located in roughly similar places in the graph and R1 to R3 follow a single trend. A single equation is proposed to parameterize SSA from the density for all four subtypes.
Figure 5. SSA-density correlation for aged snows. All subtypes follow different trends.

an air temperature above 0°C, we recommend using the average value for the F4 type: 503 ± 112 cm² g⁻¹.

Figure 3 shows that there is a trend, although fairly weak, between SSA and density for fresh snow. The same trend seems to apply to snow at T > 0°C and at T < 0°C. It also applies to recently windblown snow. As in Figure 1 and for data discussed later, several empirical equations were tested to fit the data. Overall, equations of the form SSA = A ln(d) + B had the best correlation coefficients. If the density of fresh or recently wind-blown snow is known, we propose to parameterize its SSA in models as

$$\text{SSA}_{\text{FW}} = -174.13 \ln(d) + 306.4,$$  \hspace{1cm} (3)

with SSA in cm² g⁻¹ and d in g cm⁻³. The correlation coefficient is $R^2_{\text{FW}} = 0.165$ and the mean error $E_{\text{mean}}$ made using equation (3) to predict SSA is 30%.

Likewise, Figure 4 shows that all recent snows seem to follow a single trend, and we propose that the SSA of snows of the R type be parameterized as

$$\text{SSA}_{\text{R}} = -160.51 \ln(d) + 70.1.$$  \hspace{1cm} (4)

The correlation coefficient is $R^2_{\text{R}} = 0.262$ and the mean error made using equation (4) to predict SSA is 31%. If data from subtypes R1 and R3 are fitted separately, we obtain $R^2_{\text{R1}} = 0.501$ and $R^2_{\text{R3}} = 0.364$. However, since we recommend using a single equation for the F subtypes, there is little point in proposing different equations for the R subtypes.

Aged snow subtypes, however, definitely follow different trends, as visible in Figure 5. Figure 6 shows the SSA-density correlation for the A1 subtype (rounded grains). The equation for all A1 data, with $R^2_{\text{A1}} = 0.182$ and $E_{\text{mean}} = 31\%$ is

$$\text{SSA}_{\text{A1}} = -102.31 \ln(d) + 88.9.$$  \hspace{1cm} (5)

However, Figure 6 shows that three different trends can be identified for tundra windpacks, Alpine snowpacks and the snow that was allowed to accumulate on tables in Alaska. We suggest that this is related to wind action. In the Arctic, the tundra snowpack is largely shaped by wind whose speed can exceed 30 m s⁻¹ [e.g., Dominé et al., 2002], snow crystals are broken up and sublimate before deposition, leading to hard windpacks with small grains, and thus of fairly high SSA. Their low permeability [Albert et al., 2002] impedes water vapor transport and grain growth, so that we speculate that the rate of decrease of their SSA is low. In the French Alps, wind is present but its action is reduced compared to the Arctic because storms are less violent and accumulation is much greater. Some layers are exposed to wind, while others are never affected, which accounts for a greater range of SSA values. The snow is less dense, and we speculate that the greater permeability of the snowpack allows higher rates of SSA decrease. The snow that accumulated on tables in Alaska was protected from wind action by plastic sheets on the side. Wind speed was almost always very low (<3 m s⁻¹) during these experiments anyway [Taillandier et al., 2006] (see also http://climate.gi.alaska.edu). Thus the interest of these experiments is that they represent the lower limit of wind action in the formation and evolution of layers with rounded grains. We propose the following equations:

$$\text{SSA}_{\text{A1, tundra}} = -223.53 \ln(d) + 0.6$$  \hspace{1cm} (6)

$$\text{SSA}_{\text{A1, Alpine}} = -313.17 \ln(d) + 160.1.$$  \hspace{1cm} (7)

These equations have $R^2$ values of 0.562 and 0.742 and $E_{\text{mean}}$ values of 22% and 20% respectively. Regarding the Alaska table experiments, there are too few data to propose a correlation. The most reasonable proposition with the data available is probably the average value, 155 cm² g⁻¹.
independent of density, resulting in $E_{\text{mean}} = 12\%$. An ideal parameterization of the SSA of the A1 subtype should include a wind action parameter, and we speculate that for each wind speed, a different curve would be obtained in Figure 6. The curve would be lower for lower wind speed values; that is, SSA values are increased by higher wind speeds. Although it would clearly be preferable to know the wind speed under which each layer formed to evaluate its SSA, the statistical nature of our approach does not make this an obligation. All that is needed is to know the snowpack type, i.e., to make a cursory observation of the snowpack or to have the knowledge of basic climate data, as snowpack types are determined by climate [Sturm et al., 1995]. In the absence of details on wind conditions, we recommend the use of equation (5). For tundra windpacks, we suggest testing equation (6), but the user is warned that we feel more data are needed to reach a level of confidence we consider as satisfactory. The same applies for the Alpine snowpacks and equation (7).

To a first approximation, all data within the A2 subtype follow the same trend (Figure 7), although it is clear that in the taiga snowpack, samples have a lower density and a wider range of SSA. The correlation is, with $R_{A2}^2 = 0.594$ and $E_{\text{mean}} = 27\%$:

$$\text{SSA}_{A2} = -150.67 \ln(d) - 73.5. \quad (8)$$

A closer examination shows that two different trends can actually be detected. The taiga samples appear to have their own trend. The correlation is, with $R_{A2,\text{taiga}}^2 = 0.692$ and $E_{\text{mean}} = 21\%$,

$$\text{SSA}_{A2,\text{taiga}} = -345.4 \ln(d) - 457.2. \quad (9)$$

The second trend regroups the Alpine and tundra snowpacks, as well as samples from the Alaska tables. The correlation is, with $R_{A2,\text{alp,tab}}^2 = 0.663$ and $E_{\text{mean}} = 16\%$,

$$\text{SSA}_{A2,\text{alp,tab}} = -101.0 \ln(d) + 0.9. \quad (10)$$

The reason why there are two different trends in the A2 type is probably because the temperature gradient is much greater in the taiga snowpack [Sturm and Benson, 1997]. These temperature gradients generate enormous upward water vapor fluxes that reduce the density of the lower layers of the taiga snowpack, whose density stays below 0.20 [Sturm and Benson, 1997; Taillandier et al., 2007]. After 2 to 3 weeks, faceted crystals (A2 type) of the taiga snowpack transform into depth hoar. On the contrary, in the Alpine and tundra snowpacks, the snow is much less permeable, temperature gradients are usually significantly lower, so that the large water vapor fluxes necessary to maintain a low density or to decrease it are usually not there. Hence taiga samples are all at low density ($d < 0.20$) while all but one of the Alpine and tundra samples have higher density ($d > 0.20$). The Alaska table samples were left to accumulate without any wind compaction at all, and the snowpack being thin [Sturm and Benson, 1997; Taillandier et al., 2006], the load did not lead to much compaction. Faceted crystals did form on the tables because of transient gradients in late winter, when the sun started heating surface snow. Interestingly, if the data from the tables are removed, the SSA values predicted by equation (10) change by less than 1%, confirming that those tables data do follow the Alpine and tundra trend very closely.

In summary, from our data we propose to apply equation (9) for faceted crystals having high temperature gradients generating large upward vapor fluxes. These faceted crystals are bound to transform into depth hoar rapidly, within three weeks at the most. We suggest using equation (10) for snowpack where the temperature gradient will not generate the large water vapor fluxes mentioned...
above. These faceted crystals in general will never transform into depth hoar.

[36] Depth hoar data (A3 subtype) are reported in Figure 8, and show a lot of scatter, leading to a correlation $R_{A3}^2 = 0.169$ and $E_{mean} = 28\%$ for the equation

$$SSA_{A3} = -68.862 \ln(d) + 1.2.$$  \hspace{1cm} (11)

[37] Again, the taiga data alone show a well defined trend, with $R_{A3, taiga}^2 = 0.658$ and $E_{mean} = 23\%$ for the equation

$$SSA_{A3, taiga} = -206.48 \ln(d) - 241.9,$$  \hspace{1cm} (12)

while the Alpine and tundra data show little correlation between SSA and density, with $R_{A3, alp.tun}^2 = 0.032$ and $E_{mean} = 20\%$, the equation being

$$SSA_{A3, alp.tun} = -23.97 \ln(d) + 87.2.$$  \hspace{1cm} (13)

[38] Given the poor correlation, one might as well use the average value $SSA_{A3, alp.tun} = 135$ cm$^2$g$^{-1}$, resulting in $E_{mean} = 23\%$. We also note that the lowest 2 values (87 and 91) are the only Alpine data for this class: depth hoar is not common in the French Alps. In many places, it does not form every year.

[39] We again attempt to understand why there are 2 distinct trends for depth hoar. In taiga snowpacks, depth hoar forms directly from the rapid metamorphism of precipitation under high temperature gradients. In vegetated areas such as the taiga, wind has little effect on snow compaction. This, combined with the high water vapor flux mentioned above, prevents snowpack densification. As a result, the ultimate stage of the taiga snowpack is what we may call terminal depth hoar, which has a density around 0.2 g cm$^{-3}$ and a SSA of about 75 cm$^2$ g$^{-1}$.

[40] In the tundra and Alpine snowpacks, depth hoar formation usually goes through the intermediate stage of a windpack, whose density is often in the range 0.35 to 0.5 g cm$^{-3}$. Depth hoar formation is then usually associated with mass loss by sublimation from the warmer snow layer to the colder atmosphere in the fall [e.g., Dominé et al., 2002, and references therein]. Depth hoar does not need a low-density snowpack to form, as suggested by Marbouty [1980], who proposed an upper limit to snow density of 0.35 for depth hoar formation. Near Barrow, Alaska, we have observed depth hoar that had formed in very dense basal ice layers, in frozen percolation channels, and in hard windpacks. Depth hoar density can reach 0.4 in the tundra snowpack. In the accumulation zone of Alpine glaciers, we have also observed melt layers from the previous season transformed into depth hoar of density greater than 0.4, in late fall. In summary, because the tundra and Alpine depth hoar goes through the intermediate stage of a dense layer, and because dense hoar can form in dense snow given sufficiently high temperature gradients (greater than 100°C/m), tundra and Alpine dense hoar usually have a density greater than 0.2, while taiga depth hoar has a density less than 0.22. Hence two distinct trends exist in density-SSA correlations for depth hoar. The terminal SSA value for depth hoar appears to be around 75 cm$^2$ g$^{-1}$ in the taiga, and 100 cm$^2$ g$^{-1}$ in the tundra. We only have two Alpine values and cannot draw conclusions for this snowpack type.

[41] Data for aged snow having undergone light melting, A4, are reported in Figure 9. All data are from the Alps, except one sample from the taiga snowpack in Alaska.
Given the available data, we can only propose the existence of one trend, with $R^2_{A4} = 0.578$ and $E_{mean} = 27\%$ for the equation

$$SSA_{A4} = -217.3 \ln(d) - 101.4.$$

There is significant scatter around the fit, in part caused by variable degrees of melting and the occurrence or absence of percolation.

We only have five SSA values for the A5 type (extensive melting leading to the formation of a solid icerike layer, when frozen), four of them from the Alaska taiga. Two of those are thin surface melt layers that formed in early fall. They had SSAs of 20 and 38 cm$^2$ g$^{-1}$ and their density could not be measured. The other two are from the late stages of melting of the snowpack in April. They had SSAs of 19 and 23 cm$^2$ g$^{-1}$ and densities around 0.12. The fifth value is from the Alps, with SSA = 45 cm$^2$ g$^{-1}$ and
\[ d = 0.23. \] On the basis of this limited data set, we recommend, with \( E_{\text{mean}} = 35\% \), the average value

\[
\text{SSA}_{A5} = 29 \text{ cm}^2 \text{ g}^{-1},
\]

while we do realize that the degree of melting and percolation, as well as the density, could affect SSA. However, we expect the SSA of A5 samples to be always low, and in any case much lower than 100 \( \text{cm}^2 \text{ g}^{-1} \).

\[ [44] \] Regarding the S1 type (surface hoar) we have 3 values from the Arctic (564, 336, 289 \( \text{cm}^2 \text{ g}^{-1} \)), two values from the Alps (435, 291 \( \text{cm}^2 \text{ g}^{-1} \)) and 3 values from the Alaska taiga (295, 279, 241 \( \text{cm}^2 \text{ g}^{-1} \)), but only one density value, 0.10 \( \text{g cm}^{-3} \) for the Alaska sample with SSA of 295 \( \text{cm}^2 \text{ g}^{-1} \). This measurement was performed by placing a board on the surface of the snow prior to surface hoar formation. Hence we cannot propose a SSA-density correlation for surface hoar, and only recommend, with \( E_{\text{mean}} = 26\% \), the average value

\[
\text{SSA}_{S1} = 341 \text{ cm}^2 \text{ g}^{-1}.
\]

4. Summary and Future Work

\[ [45] \] We report here 345 SSA values and show in Figure 1 that there is a correlation between SSA and density, described by equation (1). However, given the scatter around equation (1), and the clear dependence of SSA on snow type, we feel that using equation (1) to predict snow SSA in snow models is not satisfactory, and we propose three types of parameterizations that take into account snow type, as summed up in Table 2.

\[ [46] \] The most basic parameterization is based on Table 1 and Figure 2, where only the average SSA value for each snow subtype is used, independently of density. The intermediate version uses only one correlation for each snow subtype. The most sophisticated version is different from the previous one in its treatment of some aged snows only, i.e., for the A1 (rounded grains), A2 (faceted crystals) and A3 (depth hoar) subtypes. To sum up the differences between parameterizations 2 and 3, let us just say that parameterization 3 takes into account the effect of wind on the SSA-density correlations of aged snows. This effect is different for, on the one hand, the taiga snowpack where the vegetation shelters the snow from wind compaction and for, on the other hand, the tundra and Alpine snowpacks where wind densifies the snowpacks, reduces its permeability and the water vapor fluxes through it.

\[ [47] \] Despite the 345 SSA measurements presented here, these parameterizations still suffer from numerous weaknesses. First of all, it would be desirable to predict the SSA of fresh precipitated snow from the meteorological and atmospheric conditions that generated it: temperature and moisture content of the cloud, concentration of ice nuclei, rate of cooling, and vertical temperature and humidity profiles in the atmosphere. Understanding these issues is a distant goal, as most meteorological models do not have a detailed parameterization of ice nuclei.

\[ [48] \] Second, we only have one measurement of the density of surface hoar (type S1). Under cold conditions (\( T < -35^\circ \text{C} \)), such as those prevailing in the Arctic, surface hoar forms such thin layers that density measurements appear illusory. We believe that it would be more useful to predict the mass of surface hoar forming, and to study the effect of temperature and growth rate on SSA.

\[ [49] \] Third, we have few data on wet snow. The fresh (F type) and recent (R type) wet snows studied all had low degrees of melting that did not really affect the aspect of crystals during field observations. Measurements of snow more affected by melting are desirable. We also clearly have too few studies of ripening snowpacks, and we do need to perform measurements on dense snow.

\[ [50] \] Finally, the faceted crystals (A2 subtype) of the tundra snowpacks that we studied are all spring layers that were only lightly compacted by weak winds and that underwent gradient metamorphism near the surface of the snowpack. These snows did not turn into depth hoar during 24 days of monitoring in the spring, and their behavior may be different from that of the faceted crystals that precede

---

Table 2. Three Proposed Parameterizations of Snow SSA, With Increasing Sophistication

<table>
<thead>
<tr>
<th>Snow Type</th>
<th>Parameterization 1</th>
<th>Parameterization 2</th>
<th>Parameterization 3</th>
</tr>
</thead>
<tbody>
<tr>
<td>F1</td>
<td>849</td>
<td>( \text{SSA}_{A5} = -174.1 \ln(d) + 306.4 )</td>
<td>( \text{SSA}_{A5} = -174.1 \ln(d) + 306.4 )</td>
</tr>
<tr>
<td>F2</td>
<td>809</td>
<td>( \text{SSA}_{A5} = -174.1 \ln(d) + 306.4 )</td>
<td>( \text{SSA}_{A5} = -174.1 \ln(d) + 306.4 )</td>
</tr>
<tr>
<td>F3</td>
<td>658</td>
<td>( \text{SSA}_{A5} = -174.1 \ln(d) + 306.4 )</td>
<td>( \text{SSA}_{A5} = -174.1 \ln(d) + 306.4 )</td>
</tr>
<tr>
<td>F4</td>
<td>503</td>
<td>( \text{SSA}_{A5} = -174.1 \ln(d) + 306.4 )</td>
<td>( \text{SSA}_{A5} = -174.1 \ln(d) + 306.4 )</td>
</tr>
<tr>
<td>R1</td>
<td>396</td>
<td>( \text{SSA}_{A5} = -160.5 \ln(d) + 70.1 )</td>
<td>( \text{SSA}_{A5} = -160.5 \ln(d) + 70.1 )</td>
</tr>
<tr>
<td>R2</td>
<td>469</td>
<td>( \text{SSA}_{A5} = -160.5 \ln(d) + 70.1 )</td>
<td>( \text{SSA}_{A5} = -160.5 \ln(d) + 70.1 )</td>
</tr>
<tr>
<td>R3</td>
<td>356</td>
<td>( \text{SSA}_{A5} = -160.5 \ln(d) + 70.1 )</td>
<td>( \text{SSA}_{A5} = -160.5 \ln(d) + 70.1 )</td>
</tr>
<tr>
<td>R4</td>
<td>328</td>
<td>( \text{SSA}_{A5} = -160.5 \ln(d) + 70.1 )</td>
<td>( \text{SSA}_{A5} = -160.5 \ln(d) + 70.1 )</td>
</tr>
<tr>
<td>A1</td>
<td>206</td>
<td>( \text{SSA}_{A5} = -102.3 \ln(d) + 88.9 )</td>
<td>( \text{SSA}_{A5} = -102.3 \ln(d) + 88.9 )</td>
</tr>
<tr>
<td>A2</td>
<td>176</td>
<td>( \text{SSA}_{A5} = -150.7 \ln(d) - 73.5 )</td>
<td>( \text{SSA}_{A5} = -150.7 \ln(d) - 73.5 )</td>
</tr>
<tr>
<td>A3</td>
<td>120</td>
<td>( \text{SSA}_{A5} = -68.86 \ln(d) + 1.2 )</td>
<td>( \text{SSA}_{A5} = -68.86 \ln(d) + 1.2 )</td>
</tr>
<tr>
<td>A4</td>
<td>187</td>
<td>( \text{SSA}_{A5} = -217.3 \ln(d) - 101.4 )</td>
<td>( \text{SSA}_{A5} = -217.3 \ln(d) - 101.4 )</td>
</tr>
<tr>
<td>A5</td>
<td>29</td>
<td>29</td>
<td>29</td>
</tr>
<tr>
<td>S1</td>
<td>341</td>
<td>341</td>
<td>341</td>
</tr>
<tr>
<td>W1</td>
<td>604</td>
<td>( \text{SSA}_{A5} = -174.1 \ln(d) + 306.4 )</td>
<td>( \text{SSA}_{A5} = -174.1 \ln(d) + 306.4 )</td>
</tr>
</tbody>
</table>

*aSSA is in cm\(^2\) g\(^{-1}\) and snow density, \( d \), is in g cm\(^{-3}\).*
depth hoar formation at the base of the tundra snowpack in the fall. Fall measurement campaigns in the Arctic are needed to fill that data gap.

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